1 PRODUCT AND COMPANY IDENTIFICATION

**DFO**
2000 Market Street
Philadelphia, PA 19103

**EMERGENCY PHONE NUMBERS:**
Chemtrec: (800) 424-9300 (24hrs) or (703) 527-3887
Medical: Rocky Mountain Poison Control Center
(866) 767-5089 (24Hrs)

**Information Telephone Numbers**
**Product Information**
800-245-5858
8:00 am - 5:30 pm (Eastern)

**Product Information**
**Product Name**
Forane (R) 32
**Product Synonym(s)**
R-32, HFC-32

**Chemical Family**
Hydrofluorocarbon
**Chemical Formula**
CH2F2
**Chemical Name**
difluoromethane
**EPA Reg Num**

**Product Use**
Refrigerant

2 COMPOSITION / INFORMATION ON INGREDIENTS

<table>
<thead>
<tr>
<th>Ingredient Name</th>
<th>CAS RegistryNumber</th>
<th>Typical Wt. %</th>
<th>OSHA</th>
</tr>
</thead>
<tbody>
<tr>
<td>difluoromethane (HFC-32)</td>
<td>75-10-5</td>
<td>100%</td>
<td>Y</td>
</tr>
</tbody>
</table>

The substance(s) marked with a "Y" in the OSHA column, are identified as hazardous chemicals according to the criteria of the OSHA Hazard Communication Standard (29 CFR 1910.1200)

This material is classified as hazardous under Federal OSHA regulation.

The components of this product are all on the TSCA Inventory list.

3 HAZARDS IDENTIFICATION

Emergency Overview
Colorless liquefied gas with faint ether odor.
DANGER!
FLAMMABLE LIQUID AND GAS UNDER PRESSURE, OVERHEATING OR OVERPRESSURIZING MAY CAUSE GAS RELEASE OR VIOLENT CYLINDER BURSTING. MAY DECOMPOSE ON CONTACT WITH FLAMES OR EXTREMELY HOT METAL SURFACES TO PRODUCE TOXIC AND CORROSIVE PRODUCTS. VAPOR REDUCES OXYGEN AVAILABLE FOR BREATHING AND IS HEAVIER THAN AIR. HARMFUL IF INHALED AND MAY CAUSE HEART IRREGULARITIES, UNCONSCIOUSNESS OR DEATH. LIQUID CONTACT WITH EYES OR SKIN MAY CAUSE FROSTBITE.

Potential Health Effects
Skin contact and inhalation are expected to be the primary routes of occupational exposure to this material. As with most liquefied gases, contact with the rapidly volatilizing liquid can cause frostbite to any tissue. High vapor concentrations are irritating to the eyes and respiratory tract and may result in central nervous system (CNS) effects such as headache, dizziness, drowsiness and, in severe exposure, loss of consciousness and death. The dense vapor of this material may reduce the available oxygen for breathing. Prolonged exposure to an oxygen-deficient atmosphere may be fatal. Inhalation may cause an increase in the sensitivity of the heart to adrenaline, which could result in irregular or rapid heartbeats. Medical conditions aggravated by exposure to this material include heart disease or...
4 FIRST AID MEASURES

IF IN EYES, immediately flush with plenty of water. Get medical attention if irritation persists.

IF ON SKIN, flush exposed skin with lukewarm water (not hot), or use other means to warm skin slowly. Get medical attention if frostbitten by liquid or if irritation occurs.

IF SWALLOWED, Not applicable - product is a gas at ambient temperatures.

IF INHALED, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention. Do not give adrenaline, epinephrin or similar drugs following exposure to this product.

5 FIRE FIGHTING MEASURES

Fire and Explosive Properties

Auto-Ignition Temperature      NE
Flash Point                     NA - GAS
Flammable Limits - Upper       31.0 % v/v
                                Lower     14.4 % v/v

Extinguishing Media

Use water spray, water fog, carbon dioxide, or dry chemical.

Fire Fighting Instructions

Stop the flow of gas if possible. Use water spray on person making shut-off. Fire fighters and others who may be exposed to products of combustion should wear full fire fighting turn out gear (full Bunker Gear) and self-contained breathing apparatus (pressure demand NIOSH approved or equivalent). Fire fighting equipment should be thoroughly decontaminated after use.

Fire and Explosion Hazards

May decompose on contact with flames or extremely hot metal surfaces to produce toxic and corrosive products. Liquid and gas under pressure, overheating or overpressurizing may cause gas release and/or violent cylinder bursting. Container may explode if heated due to resulting pressure rise. Some mixtures of HCFCs and/or HFCs, and air or oxygen may be combustible if pressurized and exposed to extreme heat or flame.

6 ACCIDENTAL RELEASE MEASURES

In Case of Spill or Leak

Extinguish or turn off ignition or combustion sources. Evacuate enclosed spaces until gas is dispersed. Keep upwind. Stop the leak if you can do so without risk. Disperse gas with floor level forced-air ventilation. Exhaust gas outdoors.

7 HANDLING AND STORAGE

Handling

Do not get in eyes, on skin or clothing. Avoid breathing gas. Keep container closed. Use only with adequate ventilation. Wash thoroughly after handling. Keep away from heat, sparks and flame. Empty container may contain hazardous residues. Do not reuse container. Do not cut, grind or weld on or near containers - explosion hazard. Use explosion proof equipment. Use grounding and bonding connection when transferring material to prevent static discharges, fire or explosion.
7 HANDLING AND STORAGE

Storage
Do not apply direct flame to cylinder. Do not store cylinder in direct sun or expose it to heat above 120 F. Do not drop or refill this cylinder. Keep away from heat, sparks and flames.

8 EXPOSURE CONTROLS / PERSONAL PROTECTION

Engineering Controls
Investigate engineering techniques to reduce exposures below airborne exposure limits. Provide ventilation if necessary to control exposure levels below airborne exposure limits (see below). If practical, use local mechanical exhaust ventilation at sources of air contamination such as open process equipment.

Eye / Face Protection
Where there is potential for eye contact, wear chemical goggles and have eye flushing equipment available.

Skin Protection
Wear appropriate chemical resistant protective clothing and chemical resistant gloves to prevent skin contact. Consult glove manufacturer to determine appropriate type glove material for given application. Rinse contaminated skin promptly. Wash contaminated clothing and clean protective equipment before reuse. Wash skin thoroughly after handling.

Respiratory Protection
Avoid breathing gas. When airborne exposure limits are exceeded (see below), use NIOSH approved respiratory protection equipment appropriate to the material and/or its components (full facepiece recommended). Consult respirator manufacturer to determine appropriate type equipment for a given application. Observe respirator use limitations specified by NIOSH or the manufacturer. For emergency and other conditions where exposure limit may be significantly exceeded, use an approved full face positive-pressure, self-contained breathing apparatus or positive-pressure airline with auxiliary self-contained air supply. Respiratory protection programs must comply with 29 CFR § 1910.134.

Airborne Exposure Guidelines for Ingredients

<table>
<thead>
<tr>
<th>Exposure Limit</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>WEEL TWA</td>
<td>2200 mg/m3 1000 ppm</td>
</tr>
</tbody>
</table>

-Only those components with exposure limits are printed in this section.
-Skin contact limits designated with a "Y" above have skin contact effect. Air sampling alone is insufficient to accurately quantitate exposure. Measures to prevent significant cutaneous absorption may be required.
-ACGIH Sensitizer designator with a value of "Y" above means that exposure to this material may cause allergic reactions.
-WEEL-AIHA Sensitizer designator with a value of "Y" above means that exposure to this material may cause allergic skin reactions.

9 PHYSICAL AND CHEMICAL PROPERTIES

Appearance/Odor: Colorless liquified gas with faint ether odor.

pH: NE

Specific Gravity: 0.96 @ 25 C

Vapor Pressure: 221 psia @ 21.1 C (70 F)

Vapor Density: (AIR = 1) 1.81

Melting Point: NE

Freezing Point: -213 F (-136 C)

Boiling Point: -61 F/-51.7 C

Solubility In Water: NE

Percent Volatile: 100

Molecular Weight: 52.02
10 STABILITY AND REACTIVITY

Stability
This material is chemically stable under specified conditions or storage, shipment and/or use. See HANDLING AND STORAGE section of this MSDS for specified conditions.

Incompatibility
Avoid contact with strong alkali or alkaline earth metals, finely powdered metals such as aluminum, magnesium or zinc and strong oxidizers, since they may react or accelerate decomposition.

Hazardous Decomposition Products
Thermal decomposition products could include Halogen acid (HF), Carbon monoxide, Carbon dioxide, and Carbonyl halide.

11 TOXICOLOGICAL INFORMATION

Toxicological Information
Data on this material and/or its components are summarized below.
Methane, difluoro-
Single exposure (acute) studies indicate that this material is practically non-toxic if inhaled (rat 4-hr LC50 >520,000 ppm). Inhalation of this material, followed by intravenous injection of epinephrine to simulate stress reactions, resulted in cardiac sensitization in dogs. Acute inhalation of high concentrations has produced an anesthetic effect in rats. Following repeated inhalation exposure, no adverse effects were observed in rats. No birth defects were noted in the offspring of rats or rabbits exposed by inhalation during pregnancy, even at dosages which produced significant adverse effects in the mother. No genetic changes were observed in tests using bacteria, animal cells or animals.

12 ECOLOGICAL INFORMATION

Ecotoxicological Information
No data are available.

Chemical Fate Information
Data on this material and/or its components are summarized below.
Methane, difluoro-
The log Pow for this material is 1.62 indicating a low bioconcentration factor. In a 28-day ready biodegradability closed bottle test, this material appeared to be stable.

13 DISPOSAL CONSIDERATIONS

Waste Disposal
Recover, reclaim or recycle when practical. Dispose of in accordance with federal, state and local regulations. Note: Chemical additions to, processing of, or otherwise altering this material may make this waste management information incomplete, inaccurate, or otherwise inappropriate. Furthermore, state and local waste disposal requirements may be more restrictive or otherwise different from federal laws and regulations.
14 TRANSPORT INFORMATION

<table>
<thead>
<tr>
<th>DOT Name</th>
<th>Difluoromethane</th>
</tr>
</thead>
<tbody>
<tr>
<td>DOT Technical Name</td>
<td></td>
</tr>
<tr>
<td>DOT Hazard Class</td>
<td>2.1</td>
</tr>
<tr>
<td>UN Number</td>
<td>UN 3252</td>
</tr>
<tr>
<td>DOT Packing Group</td>
<td>PG NA</td>
</tr>
<tr>
<td>RQ</td>
<td></td>
</tr>
</tbody>
</table>

15 REGULATORY INFORMATION

Hazard Categories Under Criteria of SARA Title III Rules (40 CFR Part 370)

<table>
<thead>
<tr>
<th>Immediate (Acute) Health</th>
<th>Fire</th>
<th>N</th>
</tr>
</thead>
<tbody>
<tr>
<td>Delayed (Chronic) Health</td>
<td>Reactive</td>
<td>N</td>
</tr>
<tr>
<td></td>
<td>Sudden Release of Pressure</td>
<td>Y</td>
</tr>
</tbody>
</table>

The components of this product are all on the TSCA Inventory list.

Ingredient Related Regulatory Information:

New Jersey Right to Know
This product does contain the following chemical(s), as indicated below, currently on the New Jersey Right-to-Know Substances List. difluoromethane (HFC-32)

Pennsylvania Environmental Hazard
This product does contain the following chemical(s), as indicated below, currently on the Pennsylvania Environmental Hazard List. difluoromethane (HFC-32)

Pennsylvania Right to Know
This product does contain the following chemical(s), as indicated below, currently on the Pennsylvania Hazardous Substance List. difluoromethane (HFC-32)

16 OTHER INFORMATION

Revision Information

Revision Date      11 OCT 2004  Revision Number 7
Supercedes Revision Dated 18-FEB-2004

Revision Summary

ATOFINA Chemicals, Inc. has changed its name to Arkema Inc.

Key
NE = Not Established  NA = Not Applicable  (R) = Registered Trademark
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